

## **Year 10 Science - Revision Plan PPE Summer**

Below are some suggestions for revision that you can complete over the next six weeks – Please also make sure you are using the topic lists to ensure that you have covered everything needed for your exams. Any questions please speak to your science teacher. You also have Sparxs quizzes that you can use to support your revision.

<b>Week</b>	<b>Topic Area</b>	<b>Video Link</b>	<b>Activity</b>	<b>Exam question: EXTENSION</b>
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### **WEEK 1**

<b>Working Scientifically</b>	<b>Chemistry</b>	<b>Working Scientifically</b> <b>Watch the six videos linked to the following</b> Variables Systematic errors Precision Uncertainty Random errors Repeatability and reproducibility	<a href="#">GCSE Science Revision</a> <a href="#">"Independent Variable, Dependent Variable, Control Variables"</a>									
1	<b>Chemistry</b> Chemical calculations	<a href="#">GCSE Chemistry - The Mole (Higher Tier)</a>	Practice calculations in your notes using the following link: <a href="#">1.6 Calculations involving masses.pdf</a>	<table border="1" style="float: right; margin-right: 20px;"> <tr> <th>Mass number</th> <th>Percentage abundance (%)</th> </tr> <tr> <td>20</td> <td>90.48</td> </tr> <tr> <td>21</td> <td>0.27</td> </tr> <tr> <td>22</td> <td>9.25</td> </tr> </table> <p>Calculate the relative atomic mass (<math>A_r</math>) of neon. Give your answer to 3 significant figures.</p> <p style="text-align: right;">[3 mark]</p>	Mass number	Percentage abundance (%)	20	90.48	21	0.27	22	9.25
Mass number	Percentage abundance (%)											
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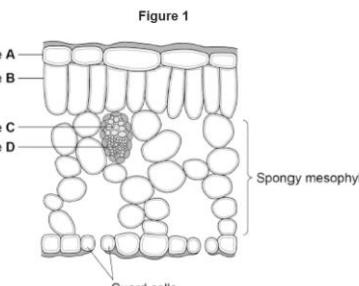
1	<p><b>Biology</b></p> <p>Specialised cells</p> <p><b>Higher</b></p> <p><u>Investigating Cells with light microscope</u></p>	<p><a href="#">GCSE Biology - Differentiation and Specialised Cells - YouTube</a></p> <p><a href="#">GCSE Science podcasts - The Cell - BBC Bitesize</a></p> <p><a href="#">GCSE Biology Revision "Required Practical 1: Microscopes"</a></p>	<p><a href="#">Cell structure - AQA test questions - GCSE Combined Science - AQA Trilogy - BBC Bitesize</a></p> <p><a href="#">Investigating cells with a light microscope - Cell structure - AQA - GCSE Combined Science Revision - AQA Trilogy - BBC Bitesize</a></p>	<p>(b) A microscope has a <math>\times 5</math> eyepiece lens.</p> <p>Describe how to use this microscope to observe a prepared slide of root hair cells at a magnification of <math>\times 50</math></p> <hr/> <hr/> <hr/> <hr/> <hr/> <hr/>
1	<p><b>Physics</b></p> <p><u>Renewable v Non renewable energy</u></p>	<p><a href="#">GCSE Physics Revision "Renewable Sources of Energy"</a></p>	<p>Make a list of Renewable sources of energy – state the pros and cons of each. Comment on reliability</p>	<p><a href="#">YEAR_9_SCIENCE_Exam_Sample_Questions_.pdf</a></p>
1	<p><b>Physics</b></p>	<p><a href="#">GCSE Physics - Wind and Solar</a></p>	<p>Think about what times of year might benefit each type and why.</p>	

## WEEK 2

2	<p><b>Chemistry</b></p> <p><b>Electrolysis</b></p>	<p><a href="#">GCSE Chemistry - Electrolysis Part 1/3 - Basics and Molten Compounds</a></p>	<p>Find the definitions for the following key words: electrolyte, anode, cathode, oxidation and reduction.</p>	<p>A mesh is used to keep the products of the electrolysis apart.</p> <p><b>[0 6] . [2]</b> Suggest <b>one</b> reason why the products of the electrolysis must be kept apart. <b>[1 mark]</b></p> <hr/> <hr/>
2	<p><b>Biology</b></p> <p>Transport in/out of cells</p> <p><b>Higher Circulatory system</b></p>	<p><a href="#">Transport in Cells: Diffusion and Osmosis   Cells   Biology   FuseSchool</a></p> <p><a href="#">Transport In Cells: Active Transport  </a></p>	<p>Describe how the following molecules get into a root hair cell:</p> <ul style="list-style-type: none"> <li>a) Oxygen</li> <li>b) Water</li> <li>c) Mineral (like potassium or magnesium)</li> </ul>	<p>(d) Arteries and veins have different structures and different functions.</p> <p>Explain how the different structure of arteries and veins relates to their different functions.</p> <hr/> <hr/> <hr/> <hr/> <hr/>

		<a href="#">Cells   Biology   FuseSchool</a>  <a href="#">GCSE Biology Revision "The Heart and Circulation"</a>	<a href="#">The heart - Animal organisation - transport systems - AQA - GCSE Combined Science Revision - AQA Trilogy - BBC Bitesize</a>	
2	<b>Physics Resistors</b>	<a href="#">GCSE Physics Revision "Resistors in Series and Parallel"</a>	Draw a series and parallel circuit with 2 resistors in each. Annotate each explaining how resistance is calculated for each circuit and general comments	<a href="#">Resistors in Series Quiz A — Isaac Physics</a>
2		<a href="#">GCSE Physics - Components</a>	Cover and test yourself on circuit symbols – name them from a diagram – draw them from name	

## WEEK 3

3	<b>Chemistry</b>  <b>Reduction and oxidation</b>	<a href="#">GCSE Chemistry - Oxidation and Reduction - Redox Reactions (Higher Tier)</a>	Recall your mnemonic for OILRIG, research which one happens at which electrode, write a paragraph explaining oxidation and reduction.	<p>Hydrogen was produced at the negative electrode and oxygen was produced at the positive electrode.</p> <p>Explain how oxygen was produced from water during the electrolysis of this aqueous solution.</p> <p style="text-align: right;">[4 marks]</p> <hr/> <hr/> <hr/> <hr/> <hr/>
3	<b>Biology</b>  Circulatory system  <b>Higher Leaf tissues</b>	<a href="#">GCSE Biology - The Circulatory System   The Heart</a>  <a href="#">GCSE Biology Revision "Plant Tissues"</a>	Click on the correct chamber or vessel in the game: <a href="#">Label the heart gcse Quiz</a>  <a href="#">Plant tissues - Plant organisation - AQA - GCSE Combined Science Revision - AQA Trilogy - BBC Bitesize</a>	<p>A leaf is a plant organ.</p> <p>Figure 1 shows tissues in a leaf.</p>  <p style="text-align: right;">Figure 1</p> <p>Label the tissues found</p>

3	<b>Physics</b> <b>National Grid</b>	<a href="#">GCSE Physics - How Transformers Work</a>	Draw and annotate a diagram of a step up transformer – Is it AC or DC?? How do the coils relate to the change in voltage?	<a href="#">Transformers Questions and Revision   MME</a>
3		<a href="#">GCSE Physics Revision "The National Grid"</a>	Create a flow chart showing how the National grid voltage changes from Power station to home	<a href="#">The National Grid Worksheets, Questions and Revision   MME</a>

## WEEK FOUR

4	<b>Chemistry</b> <b>Half equations</b>	<a href="#">GCSE Chemistry - Oxidation and Reduction - Redox Reactions (Higher Tier)</a> 2:19-end	Ensure you understand electrolysis. Recap your bonding: ionic bonding. <a href="#">Worksheet-Chemistry-Electrolysis-ks4.pdf</a>	<p><b>0 6 . 1</b> Which is the correct half equation for the production of sodium? [1 mark]</p> <p>Tick (✓) one box.</p> <p><math>\text{Na} + \text{e}^- \rightarrow \text{Na}^+</math> <input type="checkbox"/></p> <p><math>\text{Na} \rightarrow \text{Na}^+ + \text{e}^-</math> <input type="checkbox"/></p> <p><math>\text{Na}^+ + \text{e}^- \rightarrow \text{Na}</math> <input type="checkbox"/></p> <p><math>\text{Na}^+ \rightarrow \text{Na} + \text{e}^-</math> <input type="checkbox"/></p>
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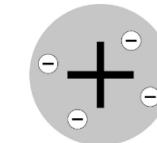
4	<b>Biology</b>  Communicable disease  <b>Higher communicable disease</b>	<a href="#">GCSE Biology - Communicable Disease</a>  <a href="#">GCSE Biology - Communicable Disease</a>	Copy and complete the table	<p>(e) Antibiotics <b>cannot</b> be used to treat measles. Suggest why.</p> <hr/> <hr/> <p>(f) Gonorrhoea is a disease caused by a bacterium. Gonorrhoea <b>can</b> be treated with antibiotics. Give <b>one</b> other way to control the spread of gonorrhoea.</p> <hr/> <hr/>
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4	<b>Data analysis</b>	<a href="#">analysis of graphs GCSE science - Google Search</a>	Watch and make notes on graph analysis	<a href="#">Graphs and Data Worksheets, Questions and Revision   MME</a>
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## WEEK FIVE

5	<b>Chemistry</b>  <b>Balancing equations</b>	<a href="#">GCSE Chemistry - Balancing Chemical Equations</a>	Ensure you know how to add in your 'big' numbers only, using a method to clearly show how you worked this out.  <a href="#">Balancing Equations</a>	<p><b>0 5 . 3</b> Complete the equation for the reaction of rubidium with water.</p> <p>You should balance the equation.</p> $\text{Rb} + \text{H}_2\text{O} \rightarrow \text{+}$
5	<b>Biology</b>  <b>Higher Photosynthesis/respiration</b>	<a href="#">GCSE Biology Revision "Photosynthesis"</a> <a href="#">GCSE Biology Revision "Respiration"</a>	<a href="#">Bioenergetics - GCSE Combined Science - BBC Bitesize</a>	<p>Light intensity varies during 24 hours.</p> <p>The figure below shows the overall exchange of carbon dioxide and oxygen at three different light intensities.</p>  <p>Bright light      Dim light      No light</p> <p>Explain the overall exchange of carbon dioxide and oxygen at the three different light intensities.</p>
5	<b>Physics</b>  <b>Radiation</b>	<a href="#">GCSE Physics - Why Radiation is Harmful</a>	Write a set of instructions to calculate the radioactivity of an unknown substance. State what equipment is needed. Include Background radiation in your answer	<a href="#">Uses and dangers of radiation - AQA test questions - GCSE Physics (Science) - AQA - BBC Bitesize</a>

## WEEK SIX

6	<b>Chemistry</b>  <b>History of an atom</b>	<a href="#">GCSE Chemistry - History of the Model of the Atom</a>	Create a fact file on the history of an atom, focusing on details of the plum pudding model.	<p>Atoms were originally thought to be tiny spheres that could not be divided.</p> <p>The plum pudding model of the atom was then developed.</p> <p>Figure 1 represents the plum pudding model of the atom.</p> <p>Figure 1</p>  <p>Describe the plum pudding model of the atom.</p> <p>[2 marks]</p>
6	<b>Biology</b>  <b>Photosynthesis</b>	<a href="#">GCSE Biology - Photosynthesis</a>	<a href="#">GCSE combined science questions - photosynthesis biology revision - BBC Bitesize</a>	

6	<b>Physics Energy</b>	<a href="#">GCSE Physics - Gravity, Weight and GPE</a>	Be confident using and rearranging this equation	<a href="#">Gravitational Potential Energy Equation - GCSE Physics</a>
		<a href="#">GCSE Physics - Internal Energy and Specific Heat Capacity</a>	Be confident using and rearranging this equation	<a href="#">YEAR_9_SCIENCE_Exam_sample_quesvions_on_SHC.p</a>
		<a href="#">GCSE Physics - Factors Affecting Gas Pressure</a>	Produce 3 flash cards on this topic	<a href="#">Gases and Pressure Questions and Revision   MME</a>

## **Triple Biology –**

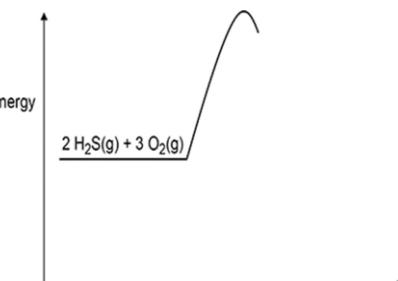
All the above, plus:

	Plant diseases	<a href="#">Plant diseases and pathogens - Higher - Plant disease - AQA - GCSE Biology (Single Science) Revision - AQA - BBC Bitesize</a>	
	Coronary heart disease	<a href="#">Development of cardiovascular disease - Animal organisation - transport systems - AQA - GCSE Combined Science Revision - AQA Trilogy - BBC Bitesize</a>	
	vaccinations	<a href="#">Vaccinations - Treating, curing and preventing disease - AQA - GCSE Combined Science Revision - AQA Trilogy - BBC Bitesize</a>	
	Aseptic technique	<a href="#">How do antiseptics affect the growth of bacteria? - Treating, curing and preventing disease - AQA - GCSE Biology (Single Science) Revision - AQA - BBC Bitesize</a>	
	Monoclonal antibodies	<a href="#">Production of monoclonal antibodies - Higher Tier - Monoclonal antibodies - Higher - AQA - GCSE Biology (Single Science) Revision - AQA - BBC Bitesize</a>	

## Triple Chemistry-

All the above, plus:

<b>5.5 Temperature change required practical</b>	<a href="#">Temperature Changes - GCSE Science Required Practical</a>	<a href="#">Exothermic &amp; Endothermic Reactions Lesson   GCSE Chemistry AQA Higher Triple   Cognito</a>	Write the required practical – include the different variables. How can you make this a fair test? Why use a polystyrene cup and not a glass one?
<b>Working Scientifically Watch the six videos linked to the following</b>  Variables Systematic errors Precision Uncertainty Random errors Repeatability and reproducibility	<a href="#">GCSE Science Revision "Independent Variable, Dependent Variable, Control Variables"</a>		
<b>5.4 - Making a soluble salt from an insoluble carbonate or oxide.</b>	<a href="#">Making Salts - GCSE Science Required Practical</a>	<a href="#">Acids &amp; Bases Lesson   GCSE Chemistry AQA Higher Triple   Cognito</a>	Write up the practical and know what the words excess mean Why do you not boil the acid? Why do you leave the crystals to dry out and not continue to heat?

<b>5.5 Calculating bond energies</b>	<a href="#">Bond Energies</a> <a href="#">Lesson   GCSE Chemistry AQA Higher Triple   Cognito</a> <p>(d) Figure 2 shows the displayed formula equation for the reaction of hydrogen sulfide with oxygen.</p> <p style="text-align: center;"><b>Figure 2</b></p> $2\text{H}-\text{S}-\text{H} + 3\text{O}=0 \longrightarrow 2\text{H}-\text{O}-\text{H} + 2\text{O}=\text{S}=0$ <p>The table below shows some of the bond energies.</p> <table border="1" data-bbox="871 446 1365 520"> <thead> <tr> <th>Bond</th><th>H—S</th><th>O=O</th><th>H—O</th><th>S=O</th></tr> </thead> <tbody> <tr> <td>Energy in kJ/mol</td><td>364</td><td>498</td><td>464</td><td>X</td></tr> </tbody> </table> <p>In the reaction the energy released forming new bonds is 1034 kJ/mol greater than the energy needed to break existing bonds.</p> <p>Calculate the bond energy X for the bond.</p> <p>Use Figure 2 and the table above.</p>	Bond	H—S	O=O	H—O	S=O	Energy in kJ/mol	364	498	464	X	<p>(c) Figure 1 shows part of the reaction profile for the reaction. The reaction is exothermic.</p> <p>Complete Figure 1.</p> <p>You should:</p> <ul style="list-style-type: none"> <li>• complete the profile line</li> <li>• label the activation energy</li> <li>• label the overall energy change.</li> </ul> <p style="text-align: center;"><b>Figure 1</b></p> 
Bond	H—S	O=O	H—O	S=O								
Energy in kJ/mol	364	498	464	X								
<b>Acid and Alkalis - Weak acids Strong acids. Titration practical</b>	<a href="#">Strong Acids &amp; Weak Acids Lesson   GCSE Chemistry AQA Higher Triple   Cognito</a> <p><b>Titration Practical Lesson   GCSE Chemistry AQA Higher Triple   Cognito</b></p>	Put together a mind map or flash cards of the differences between acids and alkalis Know the ions present How to calculate concentration in a titration.										
<b>Metals and alloys</b>	<a href="#">Metallic Bonding Lesson   GCSE Chemistry AQA Higher Triple   Cognito</a> <p><a href="#">GCSE Chemistry Revision "Alloys" (Triple)</a>  <a href="#">GCSE Chemistry Revision "Metals and Alloys"</a></p>	Draw the structure of alloys and metals What are the properties of each? Why are metals better at conducting electricity than alloys?										

[GCSE Chemistry](#)  
[Revision "Introducing](#)  
[Electrolysis"](#)

[GCSE Chemistry](#)  
[Revision "Electrolysis](#)  
[of Aqueous Solutions](#)

1"

USe the above to  
click on the next 7  
videos linking to  
electrolysis

Draw a diagram of electrolysis  
What forms at each electrode?  
What would you see?  
Write down ionic equations and examples

(c) The table below shows products of the electrolysis of two molten ionic compounds.

Complete the table.

Molten compound	Product at the negative electrode	Product at the positive electrode
Magnesium bromide	Magnesium	_____
Potassium chloride	_____	Chlorine

(2)

(d) The table below shows the products of electrolysis of two molten compounds.

Molten compound	Product at negative electrode	Product at positive electrode
Potassium iodide	Potassium	_____
Zinc bromide	_____	Bromine

Complete the table above.

(2)

## **TRIPLE Physics**

All the above, plus:

Specific latent heat	<a href="#">GCSE Physics - Specific Latent Heat</a>	Define this term and draw a graph showing the SLH of fusion and vaporisation.	<a href="#">Specific Latent Heat Questions and Revision   MME</a>
isotopes	<a href="#">Isotopes for AQA GCSE Physics and combined science</a>	Write a 1 sentence definition for an isotope and an ion.	<a href="#">Atoms, isotopes and ions - AQA test questions - GCSE Combined Science - AQA Trilogy - BBC Bitesize</a>
Half life	<a href="#">GCSE Physics Revision "Half Life"</a>	Write a set of instructions on how to calculate half life.	<a href="#">Atoms &amp; Nuclear Radiation   AQA GCSE Physics Exam Questions &amp; Answers 2016 [PDF]</a>
Irradiation v contamination	<a href="#">Contamination and irradiation - Radiation and risk - AQA Synergy - GCSE Combined Science Revision - AQA Synergy - BBC Bitesize</a>	Invent a scenario where 1 person is contaminated and 1 is irradiated – who is most at risk and why. How might being exposed to Alpha, Beta or Gamma affect the seriousness f this?	<a href="#">Irradiation vs Contamination - GCSE Physics Revision Notes</a>
GPE v KE	<a href="#">Transfer Between KE &amp; GPE Lesson   GCSE Physics AQA Higher Triple   Cognito</a>	Draw a swing in motion – explain the energy transfers as the swing slowly comes to a stop.	<a href="#">Energy Calculations - AQA GCSE Physics Revision Notes</a>
Atomic models	<a href="#">GCSE Chemistry - History of the Model of the Atom</a>	Make a timeline of the key points of how the theory of atomic structure has changed.	<a href="#">Multiple choice questions - Sample exam questions - atomic structure - AQA - GCSE Physics (Single Science) Revision - AQA - BBC Bitesize</a>
Gas Temperature and pressure	<a href="#">GCSE Physics Revision "Particle Motion in Gases"</a>	Draw a diagram of a sealed container being heated. Explain using particle theory what might happen?	<a href="#">Gases and Pressure Questions and Revision   MME</a>